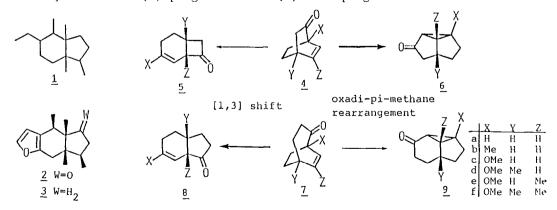
PHOTOCHEMICAL REARRANGEMENT APPROACH TO THE TOTAL SYNTHESIS OF (±)-PINGUISONE AND (±)-DEOXOPINGUISONE

Tadao Uyehara,^{*} Yasuhiro Kabasawa, Tadahiro Kato and Toshiaki Furuta (in part) Department of Chemistry, Faculty of Science, Tohoku University, Aramaki-Aoba, Sendai 980, Japan

Summary: The total synthesis of (±)-pinguisone and (±)-deoxopinguisone, the unusual [5-6] fused-ring sesquiterpenes, was accomplished by the photochemical transformation of the bicyclo[3.2.2]non-6-en-2-one into the bicyclo[4.3.0]non-4-en-7-one.

Pinguisane-type sesquiterpenes, possessing the unusual terpene skeleton $(\underline{1})$,¹⁾ have been found from various kinds of liverworts; such as pinguisone $(\underline{2})^{2}$ from <u>Aneura pinguis</u> (L.) Dum.^{2b)} and deoxopinguisone $(\underline{3})^{3}$ from <u>Ptilidium</u> ciliare (L.) Nees.^{3a)} Those compounds are of particular interest in the total synthesis of natural products, because of their novel tricyclic furan skeletons with a <u>cis</u>-junction between the five- and six-membered rings and four adjacent cis-located methyl groups.⁴⁾ In connection with our program to develop methods for stereoselective syntheses of [m-n] fused-ring terpenoids from bridged bicyclic systems, 5) we have explored the photochemical reactions of bicyclo[3.2.2]non-6-en-2-ones (7). The photochemical behavior of bicyclo[2.2.2]oct-5-en-2ones (4) has been examined extensively and the conversion into tricyclo-[3.3.0.0^{2,8}]octan-3-ones (6), via the triplet-sensitized oxadi-pi-methane rearrangement (in acetone), has been applied to syntheses of natural products.⁶⁾ On the contrary, the photochemical rearrangement of the higher homologs (7) only has a precedent in the [1,3] sigmatropic shift of 7a into 8a.⁷⁾ Herein we report the point of the photochemical reactions and the application to the total synthesis of (\pm) -pinguisone and (\pm) -deoxopinguisone.



2344

The Tieffeneau-Demjanov ring enlargement of $4b^{8}$, $4c^{9}$, and $4d^{9}$ gave expected ketones 7b¹⁰⁾, 7c, and 7d, respectively, in good yields. When treated with trimethylsilyldiazomethane in the presence of BF_3 etherate, ¹²⁾ compounds 4c, 4d, 4e, 11 and 4f underwent the ring expansion to give bicyclo[3.2.2]nonenones <u>7c-7f</u>, respectively, in more than 70% yields.

The results of the photochemical reactions of ketones 7a-7f, presented in Table 1, indicate that the preferential course is the [1,3] acyl migration to give the [5-6] cis-fused-ring system, except some reactions in acetone. The substrates which possess a methoxyl group at the C1-position did not undergo the oxadi-pi-methane rearrangement. It is interesting from synthetic viewpoints that the products derived from <u>7c-7f</u> have a masked carbonyl group in the six-membered ring and a ketone itself in the other ring.

Next, we applied this photochemical [1,3] rearrangement to the sesquiterpene synthesis outlined in Scheme 1. Bicyclo[2.2.2]octenone 11 was obtained as the major stereoisomer (ca. 10:1) in a range of 45% to 50% yields from the anisol (10) by sequential treatment with i) lithium (in ammonia), 13 ii) α -chloroacrylonitrile, and iii) potassium hydroxide and sodium sulfide.⁹⁾ That ¹H-NMR spectrum of <u>11</u> shows a doublet due to the C₈-methyl at δ =0.89, the 0.12 ppm upfield position in comparison with those of the minor stereoisomer, supports both of their stereostructures. The ring enlargement of 11¹²⁾ gave a mixture of 12 and its 3-trimethylsilyl derivative, and the mixture yielded pure $\underline{12}$ (79%) on desilylation. Photolysis of 12 in THF gave the desired ketone, 13 (59%).

After selective acetalyzation, 14 was transformed regioselectively into tetramethyl α,β -unsaturated ketone 16 (48% from 13). A dissolving metal reduction of 16 gave the alcohols, which were separeted into mixtures 17a (33%) and 17b (58%) by silica gel column chromatography. To confirm their C_q stereochemistry, each of them was converted into the corresponding ketone. The $^1\mathrm{H} ext{-NMR}$ spectrum of <u>18a</u> shows the signals for the C₉-H at δ =3.04, more than 0.5 ppm

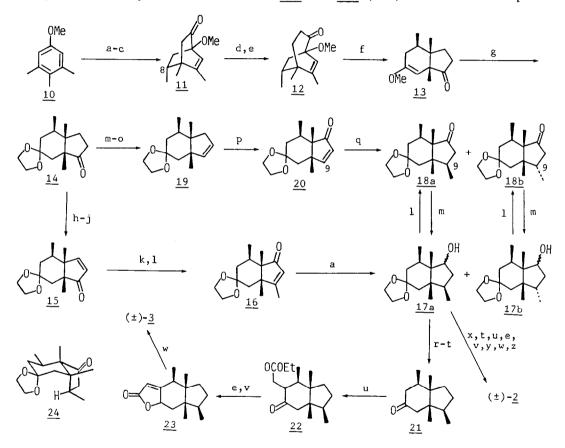
Table 1.	Photochemical rearrangements of bicyclo[3.2.2]non-6-en-2-ones"				
Substrate	Solvent	Time/h	Products, (yield, %)	Recov./%	_
<u>7a</u>	hexane	2	<u>8a</u> (53)		-
<u>7a</u>	acetone	5.5	<u>8a</u> (37), <u>9a</u> (18)	18	
<u>7</u> b	hexane	3.5	<u>8b</u> (55)		
<u>7b</u>	acetone	3.5	<u>8b</u> (28), <u>9b</u> (23)	17	
<u>7c</u>	THF	4	<u>8c</u> (55)		
<u>7d</u>	hexane	6	<u>8d</u> (56)	8	
<u>7d</u>	acetone	6	<u>8d</u> (41)		
<u>7d</u>	THF	4	<u>8d</u> (56)		
<u>7e</u>	THF	4	<u>8e</u> (47)		
<u>7f</u>	THF	4	<u>8f</u> (50)		

a) Carried out in a 0.01-0.05 M solution (ca. 100 ml) placed in an immersion well through Pyrex using a Riko 100 W high pressure mercury lamp under Ar.

a)

lower than the resonances for that of <u>18b</u>. This unusual low-field shift reflects that <u>18a</u> has the depicted configuration and conformation (<u>24</u>) in which the C_q -H is close to the C_q -axial-oxygen.

This stereochemical outcome seemed to suggest that a methyl group is introduced from the less-hindered β -side at the C₉-position of α,β -unsaturated ketone <u>20</u>. Thus we examined the other route to <u>17a</u>. Ketone <u>14</u> was reduced by use of LiAlH₄ and then the major product, the α -alcohol (82%), was converted into <u>19</u> (95%). Oxidation of <u>19</u> with chromic anhydride·3,5-dimethylpyrazole complex¹⁴) at -20 °C afforded an inseparable mixture of <u>20</u> and <u>15</u> in a ratio of 9 to 2 (79%). After treatment of this mixture with lithium dimethylcuprate at 0 °C, we could separate a mixture of <u>18a</u> and <u>18b</u> (76%) from the other products



Scheme 1. Total syntheses of (±)-pinguisone $((\pm)-\underline{2})$ and (\pm) -deoxopinguisone $((\pm)-\underline{3})^{a}$ a) (a) Li, NH₃, <u>t</u>-BuOH. (b) CH₂=C(CN)CI, PhMe, 90 °C. (c) Na₂S, KOH, EtOH, H₂O. (d) TMSCHN₂, BF₃ ether, CH₂Cl₂. (e) K₂CO₃, MeOH, H₂O. (f) $h\nu$, THF. (g) 2-ethyl-2-methyl-1,3dioxolane, TSOH, PhH. (h) LDA, PhSSO₂Ph. (i) NaIO₄, MeOH, H₂O. (j) PhMe, 90 °C. (k) MeLi, ether. (1) PCC, NaOAc, CH₂Cl₂. (m) LiAlH₄, ether. (n) MsCl, Py. (o) <u>t</u>-BuOK, DMSO. (p) CrO₃ 3,5-DMP, CH₂Cl₂. (q) (Me)₂CuLi, ether. (r) NaH, CS₂, then MeI. (s) (<u>n</u>-Bu)₃SnH, PhMe. (t) 1 M-HCl, acetone (1:4). (u) LDA, ICH₂CO₂Et, THF, HMPA. (v) TSOH, PhH. (w) DIBAH, THF, then 1 M-H₂SO₄. (x) NaH, PhCH₂Br, (<u>n</u>-Bu)₄NI, DME. (y) BBr₃, CH₂Cl₂. (z) (Ac)₂O, DMSO.

by silica gel chromatography. Lithium aluminium hydride reduction of the mixture yielded 17a and 17b in 94% and 6% yields, respectively.

In order to prepare (\pm) -deoxopinguisone, the hydroxyl group of 17a was removed¹⁵⁾ (84%) and the 1,3-dioxolane was cleaved (97%). Keto-ester <u>22</u> was prepared regioselectively (63%). Hydrolysis followed by acid-catalyzed dehydration afforded butenolide 23 in 72% yield. DIBAH reduction and subsequent acid treatment gave the furan (69%) whose spectral chracteristics are identical with those of natural deoxopinguisone.

The hydroxyl group of 17a was protected as a benzyl ether¹⁶⁾ and then the 1,3-dioxolane was cleaved (84%). The butenolide was derived from the ketone (40%) in a manner similar to the conversion of 21 into 23. After cleaving the benzvl ether¹⁷⁾ (83%), DIBAH reduction gave the stereoisomeric furan-alcohols (66%). Dimethylsulfoxide-acid anhydride oxidation¹⁸⁾ of the alcohols gave (\pm) -2 (62%), also identical spectroscopically with natural pinguisone.

We thank Professor Y. Asakawa (Tokushima Bunri University) for kindly providing a copy of the spectrum of deoxopinguisone.

References

- 1) V.Herout, "Aspects of Terpenoid Chemistry and Biochemistry," ed by T.W. Goodwin, Academic Press, London and New York (1971), p. 68.
- 2) V.Benesova, Z.Samek, V.Herout, and F.Sorm, Collect. Czeck. Chem. Commun., 34, 582 (1969); b) V.Benesova, V.Herout, and F.Sorm, <u>ibid.</u>, <u>34</u>, 1810 (1969); c) Y. Asakawa, T.Takemoto, M.Toyooka, and T.Aratani, Tetrahedron Lett., <u>1977</u>, 1407.
 3) a) S.M.Krutov, Z.Samek, V.Benesova, and V.Herout, Phytochemistry, <u>12</u>, 1405 (1973); b) V.Benesova, M.Toyota, and T.Takemoto, <u>19</u>, 251
- (1973); b) Y.Asakawa, M.Toyota, M.Kano, and T.Takemoto, ibid., 19, 2651
- (1973); b) Y.Asakawa, M.Toyota, M.Kano, and T.Takemoto, <u>1D1d</u>., <u>19</u>, 2651 (1980) and the references cited therein.
 4) S.Bernasconi, M.Ferrari, P.Gariboldi, G.Jommi, M.Sisti, and R.Destro, J. Chem. Soc., Perkin Trans. 1, <u>1981</u>, 1994; S.Bernasconi, P.Gariboldi, G. Jommi, S.Montanari, and M.Sisti, <u>ibid</u>., <u>1981</u>, 2394.
 5) T.Uyehara, K.Ogata, J.Yamada, and T.Kato, J. Chem. Soc., Chem. Commun., <u>1983</u>, 17; Bull. Chem. Soc. Jpn., <u>58</u>, 211 (1985); T.Uyehara, J.Yamada, T. Kato, and F.Bohlmann, Tetrahedron Lett., <u>24</u>, 4445 (1984).
 6) For a recent review, see: M.Demuth and K.Schaffner, Angew. Chem., Int. Ed. Engl., <u>21</u>, 820 (1982).
 7) L.A.Paquette, R.P.Henzel, and R.F.Eizember, J. Org. Chem., <u>38</u>, 3257 (1973).

- 7) L.A.Paquette, R.P.Henzel, and R.F.Eizember, J. Org. Chem., <u>38</u>, 3257 (1973).
 8) R.P.Gregson and R.N.Mirrington, Aust. J. Chem., <u>29</u>, 2037 (1976).
 9) D.A.Evans, W.L.Scott, and L.K.Truesdale, Tetrahedron Lett., <u>1972</u>, 121.
 10) Satisfactory spectral data and elemental analyses have been obtained for the key compounds described.
- 11) S.A.Monti, S.C.Chen, Y.L.Yang, S.S.Yuan, and O.P.Bourgeois, J. Org. Chem., 43, 4062 (1978).
- 43, 4002 (1970).
 12) N.Hashimoto, T.Aoyama, and T.Shioiri, Tetrahedron Lett., <u>21</u>, 4619 (1980); Chem. Pharm. Bull., <u>30</u>, 119 (1982).
 13) K.P.Dastur, J. Am. Chem. Soc., <u>96</u>, 2605 (1974).
 14) W.G.Salmond, M.A.Barta, and J.L.Havens, J. Org. Chem., <u>43</u>, 2057 (1978).
 15) D.H.R.Barton and S.W.McCombie, J. Chem. Soc., Perkin Trans. 1, <u>1975</u>, 1574.
 16) S.Czernecki, C.Georgoulis, and C.Provelenghiou, Tetrahedron Lett., <u>1976</u>, 2525

- 3535.
- 17) J.P.Kutney, N.Abdurahman, P.Le Quesne, E.Piers, and I.Vlattas, J. Am. Chem. Soc., 88, 3656 (1966).
- 18) J.D.Albright and L.Goldman, J. Am. Chem. Soc., <u>87</u>, 4214 (1965).

(Received in Japan 5 February 1985)